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Key indicators

Single-crystal X-ray study

T = 130 K

Mean $\sigma(\text{C}-\text{C}) = 0.002 \text{ \AA}$

R factor = 0.030

wR factor = 0.079

Data-to-parameter ratio = 11.8

For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.***N*-Amidinio-3,5,5-trimethyl-2-pyrazoline-
1-carboxamide chloride (BPC-151), an oral
hypoglycaemic agent**

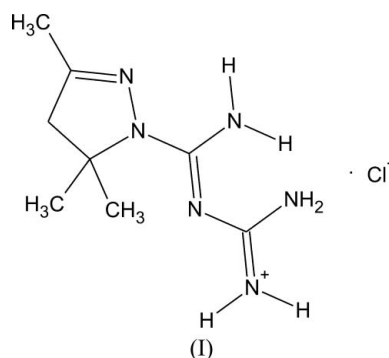
In the title compound, $\text{C}_8\text{H}_{17}\text{N}_6^+\text{Cl}^-$, the biguanidinium unit is non-planar and the bridging N atom is unprotonated. The ions are connected by hydrogen bonds into a two-dimensional network.

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Comment

Diabetes mellitus is a chronic disease that is growing in prevalence worldwide. Biguanide agents such as metformin (Campbell *et al.*, 1996) have many characteristics that are ideal for treating type 2 diabetes, including weight loss, insulin sensitization, positive lipid effect, mild hypotensive effect and low incidence of hypoglycaemia. These drugs are therefore considered to be the first-line medications. Recently, pharmacophoric features of biguanide (Bharatam *et al.*, 2005) have attracted attention due to misleading information found in the medicinal chemistry literature regarding its tautomeric structure (Glennon *et al.*, 2003; Rahman *et al.*, 2003). Previously in our laboratories, a series of biguanide derivatives have been synthesized and the title compound, (I), (BPC-151) has been selected as a useful oral hypoglycemic agent (Brzozowski *et al.*, 1979, 1981; Negwer, 1987). Here we report the crystal structure of (I) because the tautomeric structure of this compound also needs clarification.



The asymmetric unit of (I) consists of a monoprotonated cation and a chloride anion (Fig. 1). The bridging N atom of the biguanidinium fragment, N8, is not protonated, a feature typical of all biguanidinium monocations in the crystalline state (Cambridge Structural Database, Version 5.27, plus January 1st 2006 update; Allen, 2002). The biguanidinium unit is non-planar, but the C–N bond lengths, which are in the range 1.3207 (18)–1.3502 (17) Å (Table 1), suggest some degree of delocalization of π -electron density through this fragment. There is an intramolecular N11–H112 \cdots N7 interaction (Table 2), leading to pyramidalization of the N7 amino group (sum of bond angles at N7 = 351°). Intermolecular hydrogen bonds between cation and anion assemble the

component species in (I) into a strongly corrugated two-dimensional network parallel to (100), as shown in Fig. 2.

Experimental

The title compound was prepared according to a known procedure (Brzozowski *et al.*, 1979). Crystals of (I) for X-ray analysis were obtained by recrystallization from an aqueous solution.

Crystal data

$C_8H_{17}N_6^+ \cdot Cl^-$	$Z = 4$
$M_r = 232.73$	$D_x = 1.294 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 8.5463 (7) \text{ \AA}$	$\mu = 0.30 \text{ mm}^{-1}$
$b = 10.5588 (9) \text{ \AA}$	$T = 130 \text{ K}$
$c = 13.2894 (11) \text{ \AA}$	Block, colourless
$\beta = 95.150 (7)^\circ$	$0.20 \times 0.10 \times 0.10 \text{ mm}$
$V = 1194.38 (17) \text{ \AA}^3$	

Data collection

Kuma KM-4-CCD κ -geometry diffractometer	2425 independent reflections
ω scans	2047 reflections with $I > 2\sigma(I)$
Absorption correction: none	$R_{\text{int}} = 0.015$
6642 measured reflections	$\theta_{\text{max}} = 26.4^\circ$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0446P)^2 + 0.2097P]$
$R[F^2 > 2\sigma(F^2)] = 0.030$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.079$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.06$	$\Delta\rho_{\text{max}} = 0.24 \text{ e \AA}^{-3}$
2425 reflections	$\Delta\rho_{\text{min}} = -0.21 \text{ e \AA}^{-3}$
205 parameters	Extinction correction: <i>SHELXL97</i>
All H-atom parameters refined	Extinction coefficient: 0.018 (2)

Table 1

Selected bond lengths (\AA).

N2—C6	1.3502 (17)	N8—C9	1.3332 (17)
C6—N8	1.3207 (18)	C9—N10	1.3238 (19)
C6—N7	1.3448 (18)	C9—N11	1.3353 (18)

Table 2

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
N10—H102 \cdots Cl1	0.866 (19)	2.355 (19)	3.1767 (14)	158.4 (16)
N10—H101 \cdots Cl1 ⁱ	0.812 (19)	2.43 (2)	3.2271 (14)	168.8 (17)
N11—H111 \cdots Cl1	0.785 (19)	2.56 (2)	3.2965 (16)	156.3 (17)
N7—H72 \cdots Cl1 ⁱⁱ	0.90 (2)	2.29 (2)	3.1687 (13)	165.0 (16)
N11—H112 \cdots N7	0.832 (19)	2.386 (18)	2.891 (2)	119.8 (15)

Symmetry codes: (i) $-x + 2, y - \frac{1}{2}, -z + \frac{1}{2}$; (ii) $-x + 2, -y + 2, -z + 1$.

All H atoms were located in electron-density difference maps and their positional and displacement parameters included in the refinement [$C-H = 0.948 (16)$ – $1.009 (16)$; $N-H = 0.785 (19)$ – $0.866 (19)$].

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2004); cell refinement: *CrysAlis CCD*; data reduction: *CrysAlis RED* (Oxford Diffraction, 2004); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *Stereochemical Workstation Operation Manual* (Siemens, 1989) and *Mercury* (Version 1.4; Bruno *et al.*, 2002); software used to prepare material for publication: *SHELXL97*.

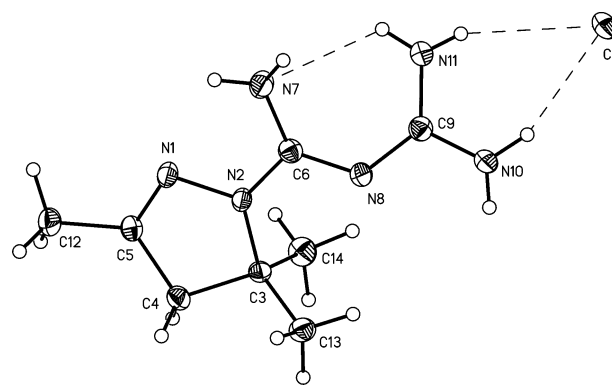


Figure 1

The molecular structure of (I), with 50% probability displacement ellipsoids. Hydrogen bonds are shown as dashed lines.

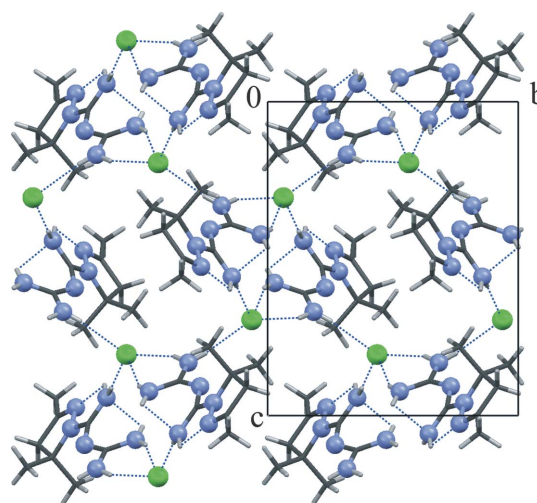


Figure 2

The two-dimensional network of hydrogen-bonded ions in (I). Hydrogen bonds are shown as dashed lines.

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